and probably on the H atoms bonded to the nitrogens as already shown by CNDO/2 calculations on imidazolines (Carpy, Léger, Leclerc, Decker, Rouot \& Wermuth, 1982).

Among the centres, those which have commonly been implicated in the interaction of $\alpha$-ligands with their receptor are the quaternary N and the hydroxyphenyl ring (Pullman, Coubeils, Courrière \& Gervois, 1972). We have calculated the distances $D$ between the two nitrogens of the guanidine function and the centre $\pi$ of the hydroxyphenyl ring: $\mathrm{N}(16)-\pi=4.40$ (1) and $\mathrm{N}(19)-\pi=5 \cdot 12(1) \AA$ and the distances $H$ between the two nitrogens and the plane containing the ring: -1.575 (3) and 0.191 (3) $\AA$ respectively. The distance $D$ related to $\mathrm{N}(19)$ is in good agreement with those found in imidazoline agonists ( $\simeq 5 \AA$ ) but $H$ is much smaller ( $\sim 0.2 \AA$ instead of $\sim 1 \AA$ ). In contrast this $H$ distance seems consistent with those found in various $\alpha$-antagonists such as WB-4101 (Carpy, Colleter \& Léger, 1981) and raubasine (Dubost, Léger, Goursolle, Carpy \& Colleter, 1983).
The crystalline cohesion is ensured by two hydrogen bonds involving the two nitrogens of the imidazoline ring and the $\mathrm{Cl}^{-}$ions: $\mathrm{N}(16) \cdots \mathrm{Cl}(22)(2-x, 1-y$, $1-z)=3.141(3) \AA, \quad \mathrm{H}(160) \cdots \mathrm{Cl}(22)=2.35(4) \AA$, $\mathrm{N}-\mathrm{H} \cdots \mathrm{Cl}=139(3)^{\circ}, \mathrm{Cl}(22) \cdots \mathrm{N}(19)\left(x, \frac{1}{2}-y, \frac{1}{2}+z\right)=$ 3.224 (3) $\AA, \mathrm{Cl}(22) \cdots \mathrm{H}(190)=2.27$ (4) $\AA, \mathrm{Cl} \cdots \mathrm{H}-\mathrm{N}$ $=167(3)^{\circ}$, by $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ bonds and by van der Waals contacts.

The sample was kindly provided by CIBA-GEIGY Limited (Basle, Switzerland).

## References

Carpy, A., Colleter, J.-C. \& Léger, J.-M. (1981). Cryst. Struct. Commun. 10, 1391-1395.
Carpy, A., Gadret, M. \& Léger, J.-M. (1979). Acta Cryst. B35, 994-996.
Carpy, A., Hickel, D. \& Léger, J.-M. (1979). Cryst. Struct. Commun. 8, 433-437.
Carpy, A., Hickel, D. \& Léger, J.-M. (1980). Cryst. Struct. Commun. 9, 43-47.
Carpy, A., Léger, J.-M., Leclerc, G., Decker, N., Rouot, B. \& Wermuth, C. G. (1982). Mol. Pharmacol. 21, 400-408.
Carpy, A., Léger, J.-M., Wermuth, C. G. \& Leclerc, G. (1981). Acta Cryst. B37, 885-889.
Dubost, J.-P., Léger, J.-M., Goursolle, M., Carpy, A. \& Colleter, J.-C. (1983). Acta Cryst. To be published.
International Tables for X-ray Crystallography (1974). Vol. IV. Birmingham: Kynoch Press.
Langer, S. Z. (1977). Br.J. Pharmacol. 60, 481-497.
Main, P., Hull, S. E., Lessinger, L,, Germain, G., Declerce, J. P. \& Woolfson, M. M. (1978). MULTAN78. A System of Computer Programs for the Automatic Solution of Crystal Structures from X-ray Diffraction Data. Univs. of York, England, and Louvain, Belgium.
Pullman, B., Coubeils, J.-L., Courrière, Ph. \& Gervois, J.-P. (1972). J. Med. Chem. 15, 17-23.

Starke, K. (1977). Rev. Physiol. Biochem. Pharmacol. 77, 1-24.
Stewart, R. F., Davidson, E. R. \& Simpson, W. T. (1965). J. Chem. Phys. 42, 3175-3187.
Westrall, T. A. (1977). Physiol. Rev. 57, 659-728.

# Structure of (+)-2,2-Dimethoxy-2-(6-methoxy-2-naphthyl)-1-methylethyl 2-Oxo-10-bornanesulfonate, $\mathrm{C}_{26} \mathrm{H}_{34} \mathrm{O}_{7} \mathrm{~S}$ 

By S. Ohba, G. Tsuchihashi and Y. Saito<br>Department of Chemistry, Faculty of Science and Technology, Keio University, Hiyoshi 3, Kohoku-ku, Yokohama 223, Japan

and K. Kitajima and S. Mitamura
Sagami Chemical Research Center, Nishi-Ohnuma, Sagamihara, Kanagawa 229, Japan
(Received 21 March 1983; accepted 14 June 1983)


#### Abstract

M_{r}=490 \cdot 6\), orthorhombic, $P 2_{1} 2_{1} 2_{1}, a=$ 10.703 (1), $\quad b=29.091$ (5) , $\quad c=8.435$ (1) $\AA$ A, $\quad V=$ $2626.4(5) \AA^{3}, \quad Z=4, \quad D_{x}=1.24 \mathrm{Mg} \mathrm{m}^{-3}, \quad \lambda(\mathrm{Cu} K \alpha)$ $=1.5418 \AA, \quad \mu=1.41 \mathrm{~mm}^{-1}, \quad F(000)=1048, \quad T=$ 298 (2) K. Final $R=0.047$ for 1916 unique observed reflections. The absolute configuration around the asymmetric C atom has been established as $R$.


0108-2701/83/101432-03\$01.50

Introduction. New methods for the synthesis of $\alpha$-arylalkanoic acids by the use of 1,2-rearrangement of the aryl group have been designed to obtain biologically important substances (Tsuchihashi, Kitajima \& Mitamura, 1981). Crystals of the title compound (I), a starting material of the 1,2 -rearrangement reaction, have been subjected to X-ray crystal-structure analysis
in order to establish the absolute configuration around the asymmetric C atom and to elucidate the reaction mechanism, especially the stereochemical course of the reaction.

(I)

Experimental. The mixture of diastereoisomers of (I) synthesized by reacting 2 -hydroxy-1-(6-methoxy-2-naphthyl)-1-propanone dimethyl acetal in pyridine with 2-oxo-10-bornanesulfonyl chloride. One of the diastereoisomers having $[\alpha]_{D}^{25}{ }^{\circ} \mathrm{C}$ of $+32.5^{\circ}\left[\mathrm{CHCl}_{3}\right.$, $10.00 \mathrm{~g} \mathrm{dm}^{-3}$ ] was isolated by recrystallization from methanol solution (m.p. 375-378 K). Thin tabular (001) crystals, approximate dimensions $0.5 \times 0.2 \times$ 0.05 mm . Preliminary determination of lattice parameters and space group ( $P 2_{1} 2_{1} 2_{1}$ from systematic absences: $h 00, h$ odd; $0 k 0, k$ odd; $00 l, l$ odd) made with a Weissenberg camera. Philips automated four-circle diffractometer, graphite monochromator, $\mathrm{Cu} K \alpha$ radiation. Cell dimensions obtained by least-squares methods from $2 \theta$ values of 15 reflections ( $38<$ $2 \theta<80^{\circ}$ ). $\theta-2 \theta$ scan technique, $2 \theta \leq 120^{\circ}(h 0-12$, $k 0-32, l 0-9)$, scan speed $2^{\circ} \mathrm{min}^{-1}(\theta)$. No significant variation in 3 standard reflections during data collection. 2267 independent reflections measured, 1916 intensities with $\left|F_{o}\right|>3 \sigma\left(\left|F_{o}\right|\right)$ considered observed and used for structure determination. Corrections for Lorentz, polarization, not for absorption. Position of S atom deduced from a Patterson map, attempts to solve structure by heavy-atom method failed; structure solved by direct methods with MULTAN78 (Main, Hull, Lessinger, Germain, Declercq \& Woolfson, 1978) and refined by block-diagonal least squares with anisotropic thermal parameters for all non-H atoms using UNICSIII computation program system (Sakurai \& Kobayashi, 1979). All H atoms located from a difference synthesis and introduced in refinement. $\sum w\left(\left|F_{g}\right|-\left|F_{c}\right|\right)^{2}$ minimized; weights assigned as $w=0.25$ for $\left|F_{l}\right| \geq 15$ and $w=1.0$ for $\left|F_{o}\right|<15$; final $R=0.047, w R=0.037, \quad S=1.0,477$ parameters refined. Maximum shift for $x$ coordinate of $\mathrm{H}(\mathrm{C} 21)$ $=2 \sigma . \Delta \rho=-0.24-0.24 \mathrm{e}^{\AA^{-3}}$. Complex neutral-atom scattering factors from International Tables for $X$-ray Crystallography (1974).

The absolute structure was determined with reference to the known absolute configuration of $(+)$-camphor (2-bornanone) (Wunderlich, 1967). In order to confirm the absolute configuration, a refinement was made separately with the enantiomeric structure. It converged to $R=0.050$ and $w R=0.039$ and could be rejected at the 0.005 significance level by the Hamilton (1965) test.

Table 1. Positional parameters ( $\times 10^{4}$; for $\mathrm{S} \times 10^{5}$ ) and equivalent isotropic temperature factors (Hamilton, 1959)

|  | $x$ | $y$ | $z$ | $B_{\text {eq }}\left(\AA^{2}\right)$ |
| :---: | :---: | :---: | :---: | :---: |
| S | 84851 (14) | 23895 (5) | -6461 (16) | 4.0 |
| O(1) | 7272 (4) | 2571 (1) | -988 (4) | $5 \cdot 0$ |
| O(2) | 9323 (4) | 2288 (1) | -1921 (4) | 5.3 |
| $\mathrm{O}(3)$ | 8208 (3) | 1932 (1) | 309 (4) | 3.4 |
| O(4) | 9723 (3) | 994 (1) | 2466 (4) | 3.9 |
| $\mathrm{O}(5)$ | 8828 (3) | 1692 (1) | 3288 (4) | $3 \cdot 3$ |
| $\mathrm{O}(6)$ | 7702 (4) | 3552 (1) | 100 (5) | $6 \cdot 1$ |
| $\mathrm{O}(7)$ | 2930 (4) | -218(1) | 449 (7) | $8 \cdot 2$ |
| C(1) | 9326 (5) | 2753 (2) | 702 (6) | $3 \cdot 6$ |
| C(2) | 8574 (5) | 3002 (2) | 1974 (6) | $3 \cdot 3$ |
| C(3) | 7919 (5) | 3440 (2) | 1440 (7) | $4 \cdot 6$ |
| C(4) | 7650 (6) | 3719 (2) | 2934 (8) | 5.7 |
| C(5) | 8234 (5) | 3412 (2) | 4245 (8) | 4.8 |
| C(6) | 7395 (6) | 2997 (2) | 4479 (8) | 4.9 |
| C(7) | 7566 (5) | 2721 (2) | 2887 (6) | 3.8 |
| C(8) | 9384 (5) | 3208 (2) | 3350 (7) | 3.7 |
| C(9) | 10071(6) | 2849 (2) | 4290 (8) | 4.7 |
| C(10) | 10334 (6) | 3566 (2) | 2754 (8) | 5.4 |
| C(11) | 9274 (7) | 1264 (2) | -833 (7) | 5.8 |
| C(12) | 9198 (5) | 1589 (2) | 570 (6) | $3 \cdot 8$ |
| C(13) | 8826 (5) | 1339 (2) | 2129 (6) | $3 \cdot 4$ |
| C(14) | 10965 (5) | 1150 (2) | 2790 (8) | $5 \cdot 1$ |
| C(15) | 8534 (6) | 1535 (2) | 4867 (6) | $4 \cdot 5$ |
| C(16) | 5326 (5) | 1152 (2) | 1910 (6) | 3.6 |
| C(17) | 6461 (5) | 1359 (2) | 2119 (6) | 3.4 |
| C(18) | 7561 (5) | 1097 (2) | 1998 (6) | 3.4 |
| C(19) | 7490 (5) | 638 (2) | 1686 (8) | $5 \cdot 0$ |
| $\mathrm{C}(20)$ | 6324 (6) | 427 (2) | 1433 (8) | $5 \cdot 1$ |
| C(21) | 6235 (7) | -48(2) | 1037 (13) | 9.3 |
| C(22) | 5093 (7) | -237(2) | 749 (13) | 9.7 |
| C(23) | 3998 (6) | 23 (2) | 807 (10) | 6.7 |
| C(24) | 4050 (6) | 476 (2) | 1202 (8) | 4.4 |
| C(25) | 5226 (5) | 682 (2) | 1519 (7) | 3.8 |
| C(26) | 1798 (6) | 23 (2) | 439 (11) | 8.4 |



Fig. 1. An ORTEP drawing of the molecule with thermal ellipsoids scaled at the $50 \%$ probability level. H atoms are represented by circles of radius $0.08 \AA$ A.

Discussion. Final atomic parameters are listed in Table 1.* An ORTEP drawing (Johnson, 1965) of the molecule is shown in Fig. 1, and bond lengths and angles are given in Table 2. The absolute configuration around the asymmetric C atom $\mathrm{C}(12)$ is $R$. Methyl (-) $\alpha$-(6-methoxy-2-naphthyl)propionate (II) was obtained by hydrolysis of (I) under reflux with an equimolar amount of calcium carbonate in a mixture of

[^0]Table 2. Bond lengths $(\AA)$ and angles $\left({ }^{\circ}\right)$

| S-O(1) | 1.431 (4) | $\mathrm{C}(2)-\mathrm{C}(3)$ | 1.522 (8) |
| :---: | :---: | :---: | :---: |
| $\mathrm{S}-\mathrm{O}(2)$ | 1.431 (4) | $\mathrm{C}(2)-\mathrm{C}(7)$ | 1.557 (8) |
| $\mathrm{S}-\mathrm{O}(3)$ | 1.584 (3) | $\mathrm{C}(2)-\mathrm{C}(8)$ | 1.568 (8) |
| $\mathrm{S}-\mathrm{C}(1)$ | 1.795 (6) | $\mathrm{C}(3)-\mathrm{C}(4)$ | 1.526 (9) |
| $\mathrm{O}(3)-\mathrm{C}(12)$ | 1.472 (6) | $\mathrm{C}(4)-\mathrm{C}(5)$ | 1.553 (9) |
| $\mathrm{O}(4)-\mathrm{C}(13)$ | 1.418 (6) | C(5)-C(6) | 1.518 (8) |
| $\mathrm{O}(4)-\mathrm{C}(14)$ | 1.431 (6) | $\mathrm{C}(5)-\mathrm{C}(8)$ | 1.561 (8) |
| $\mathrm{O}(5)-\mathrm{C}(13)$ | 1.418 (6) | $\mathrm{C}(6)-\mathrm{C}(7)$ | 1.575 (8) |
| $\mathrm{O}(5)-\mathrm{C}(15)$ | 1.443 (6) | $\mathrm{C}(8)-\mathrm{C}(9)$ | 1.503 (8) |
| $\mathrm{O}(6)-\mathrm{C}(3)$ | 1.199 (7) | $\mathrm{C}(8)-\mathrm{C}(10)$ | 1.540 (8) |
| $\mathrm{O}(7)-\mathrm{C}(23)$ | 1.375 (8) | $\mathrm{C}(11)-\mathrm{C}(12)$ | 1.517 (8) |
| O (7)-C(26) | 1.400 (7) | $\mathrm{C}(12)-\mathrm{C}(13)$ | 1.555 (7) |
| $\mathrm{C}(1)-\mathrm{C}(2)$ | 1.524 (8) | $\mathrm{C}(13)-\mathrm{C}(18)$ | 1.530 (8) |
| $\mathrm{O}(1)-\mathrm{S}-\mathrm{O}(2)$ | 119.6 (2) | $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(8)$ | 101.0 (5) |
| $\mathrm{O}(1)-\mathrm{S}-\mathrm{O}(3)$ | 104.0 (2) | $\mathrm{C}(6)-\mathrm{C}(5)-\mathrm{C}(8)$ | 103.1 (5) |
| $\mathrm{O}(1)-5-\mathrm{C}(1)$ | 111.4 (3) | $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(7)$ | 103.1 (5) |
| $\mathrm{O}(2)-\mathrm{S}-\mathrm{O}(3)$ | 109.0 (2) | $\mathrm{C}(2)-\mathrm{C}(7)-\mathrm{C}(6)$ | 103.6 (4) |
| $\mathrm{O}(2)-\mathrm{S}-\mathrm{C}(1)$ | 106.5 (2) | $\mathrm{C}(2)-\mathrm{C}(8)-\mathrm{C}(5)$ | 93.9 (4) |
| $\mathrm{O}(3)-\mathrm{S}-\mathrm{C}(1)$ | 105.5 (2) | $\mathrm{C}(2)-\mathrm{C}(8)-\mathrm{C}(9)$ | 113.3 (5) |
| $\mathrm{S}-\mathrm{O}(3)-\mathrm{C}(12)$ | 120.7 (3) | $\mathrm{C}(2)-\mathrm{C}(8)-\mathrm{C}(10)$ | 112.5 (5) |
| $\mathrm{C}(13)-\mathrm{O}(4)-\mathrm{C}(14)$ | 116.3 (4) | $\mathrm{C}(5)-\mathrm{C}(8)-\mathrm{C}(9)$ | 113.2 (5) |
| $\mathrm{C}(13)-\mathrm{O}(5)-\mathrm{C}(15)$ | 114.0 (4) | $\mathrm{C}(5)-\mathrm{C}(8)-\mathrm{C}(10)$ | 114.9 (5) |
| $\mathrm{C}(23)-\mathrm{O}(7)-\mathrm{C}(26)$ | 117.8 (4) | $\mathrm{C}(9)-\mathrm{C}(8)-\mathrm{C}(10)$ | 108.6 (5) |
| $\mathrm{S}-\mathrm{C}(1)-\mathrm{C}(2)$ | 117.5 (4) | $\mathrm{O}(3)-\mathrm{C}(12)-\mathrm{C}(11)$ | 110.1 (4) |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | 115.6 (4) | $\mathrm{O}(3)-\mathrm{C}(12)-\mathrm{C}(13)$ | 105.0 (4) |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(7)$ | 117.7 (5) | $\mathrm{C}(11)-\mathrm{C}(12)-\mathrm{C}(13)$ | 112.5 (5) |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(8)$ | 114.3 (4) | $\mathrm{O}(4)-\mathrm{C}(13)-\mathrm{O}(5)$ | 111.9 (4) |
| $\mathrm{C}(3)-\mathrm{C}(2)-\mathrm{C}(7)$ | 105.5 (4) | $\mathrm{O}(4)-\mathrm{C}(13)-\mathrm{C}(12)$ | 109.1 (4) |
| $\mathrm{C}(3)-\mathrm{C}(2)-\mathrm{C}(8)$ | 98.8 (4) | $\mathrm{O}(4)-\mathrm{C}(13)-\mathrm{C}(18)$ | 106.7 (4) |
| $\mathrm{C}(7)-\mathrm{C}(2)-\mathrm{C}(8)$ | 102.6 (4) | $\mathrm{O}(5)-\mathrm{C}(13)-\mathrm{C}(12)$ | 104.1 (4) |
| $\mathrm{O}(6)-\mathrm{C}(3)-\mathrm{C}(2)$ | 126.5 (5) | $\mathrm{O}(5)-\mathrm{C}(13)-\mathrm{C}(18)$ | 112.6 (4) |
| $\mathrm{O}(6)-\mathrm{C}(3)-\mathrm{C}(4)$ | 126.7 (5) | $\mathrm{C}(12)-\mathrm{C}(13)-\mathrm{C}(18)$ | 112.4 (4) |
| $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(4)$ | 106.7 (5) | $\mathrm{O}(7)-\mathrm{C}(23)-\mathrm{C}(22)$ | 114.5 (5) |
| $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(5)$ | 101.9 (5) | $\mathrm{O}(7)-\mathrm{C}(23)-\mathrm{C}(24)$ | 125.6 (6) |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)$ | 108.2 (5) |  |  |

water and $N, N$-dimethylformamide. 1,2-rearrangement of the 6 -methoxy-2-naphthyl group took place during the course of the hydrolysis. The absolute configuration around the asymmetric C atom of (II) could be deduced as $R$ based on the known absolute configuration of $(+)-(R)-\alpha$-(6-methoxy-2-naphthyl)propionic acid (Kitajima, Mitamura, Kobayashi, Tsuchihashi, Ohba \& Saito, 1982). Thus it was revealed that the inversion of the absolute configuration around the asymmetric C atom occurs with the aryl-group rearrangement.

The torsion angle $\mathrm{O}(3)-\mathrm{C}(12)-\mathrm{C}(13)-\mathrm{C}(18)$ is $60.9(5)^{\circ}$. The two bulky groups of 6 -methoxy-2naphthyl $[\mathrm{C}(18)]$ and 2-oxo-10-bornanesulfonate $[\mathrm{O}(3)]$ are in a gauche position with respect to the $C(12)-C(13)$ bond axis. However, there is no short contact between these groups. The two methoxy groups bonded to the $\mathrm{C}(13)$ atom are bent into a spiral shape,
the dihedral angle between the $\mathrm{C}(13)-\mathrm{O}(5)-\mathrm{C}(15)$ and $\mathrm{C}(13)-\mathrm{O}(4)-\mathrm{C}(14)$ planes being $87(1)^{\circ}$. Four intramolecular distances less than $2.9 \AA$ are observed for the O atoms of these methoxy groups. The values of $O(4) \cdots C(15), \quad O(4) \cdots C(19), \quad O(5) \cdots O(3)$ and $O(5) \cdots C(14)$ are 2.863 (6), 2.687 (6), 2.691 (5) and 2.810 (6) $\AA$, respectiveiy. The naphthyl moiety is essentially planar with deviations less than $0.05 \AA$ for each atom in the least-squares plane. The torsion angle $\mathrm{C}(24)-\mathrm{C}(23)-\mathrm{O}(7)-\mathrm{C}(26)$ is $2(1)^{\circ}$. The bond angle $\mathrm{C}(24)-\mathrm{C}(23)-\mathrm{O}(7)$ of $125.6(6)^{\circ}$ is greater than $\mathrm{C}(22)-\mathrm{C}(23)-\mathrm{O}(7)$ of $114.5(5)^{\circ}$ because of the steric interaction between $C(24)$ and $C(26)[2.821$ (9) $\AA$ ].

The crystal structure consists of discrete molecules separated by normal van der Waals distances.* The shortest intermolecular distance is $3 \cdot 185(8) \AA$ for $\mathrm{O}(6) \cdots \mathrm{C}(14)\left(-\frac{1}{2}+x, \frac{1}{2}-y,-z\right)$.

The authors are indebted to the Institute for Solid State Physics, The University of Tokyo for the use of a FACOM M-160F computer. Part of the cost of this investigation was met by a Scientific Research Grant from The Ministry of Education to which the authors' thanks are due.

* See deposition footnote.


## References

Hamilton, W. C. (1959). Acta Cryst. 12, 609-610.
Hamilton, W. C. (1965). Acta Cryst. 18, 502-510.
International Tables for X-ray Crystallography (1974). Vol. IV. Birmingham: Kynoch Press.
Johnson, C. K. (1965). ORTEP. Report ORNL-3794. Oak Ridge National Laboratory, Tennessee.
Kitajima, K., Mitamura, S., Kobayashi, K., Tsuchihashi, G., Ohba, S. \& Saito, Y. (1982). 45th Annu. Meet. Chem. Soc. Jpn, Tokyo.
Main, P., Hull, S. E., Lessinger, L., Germain, G., Declerce, J.-P. \& Woolfson, M. M. (1978). MULTAN78. A System of Computer Programs for the Automatic Solution of Crystal Structures from X-ray Diffraction Data. Univs. of York, England, and Louvain, Belgium.
Sakural, T. \& Kobayashi, K. (1979). Rikagaku Kenkyusho Hokoku, 55, 69-77 (in Japanese).
Tsuchihashi, G., Kitajima, K. \& Mitamura. S. (1981). Tetrahedron Lett. 22, 4305-4308.
Wunderlich, J. A. (1967). Acta Cryst. 23, 846-855.

# The Structure of $N, N^{\prime}$-Hexamethylenedibenzamide, $\mathrm{C}_{20} \mathbf{H}_{24} \mathbf{N}_{2} \mathrm{O}_{2}$ 

## By Claude Pineault and François Brisse

Département de Chimie, Université de Montréal, CP 6210, Succ. A, Montréal, Québec, Canada H3C 3V1
(Received 24 March 1983; accepted 1 June 1983)


#### Abstract

M_{r}=324.43\), monoclinic, $\quad P 2_{1} / c, \quad a=1.247 \mathrm{Mg} \mathrm{m}^{-3}, \quad \lambda(\mathrm{Cu} K \bar{\alpha})=1.54178 \AA, \quad \mu(\mathrm{Cu} K \alpha)=$ $5.136(2), \quad b=5.346(1), \quad c=31.497(12), \quad \beta=0.61 \mathrm{~mm}^{-1}, F(000)=348, T=295 \mathrm{~K}, R=0.044$ for $92.81(3)^{\circ}, \quad V=863.8 \AA^{3}, Z=2, \quad D_{m}=1.23, D_{x}=1219$ observed reflections. The structure of $N, N^{\prime}$ -0108-2701/83/101434-04\$01.50 (c) 1983 International Union of Crystallography


[^0]:    * Lists of structure factors, anisotropic thermal parameters. mean-square displacement tensors. H-atom coordinates. bond lengths and bond angles in the naphthyl ring and a projection of the crystal structure along c have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 38674 (14 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography. 5 Abbey Square. Chester CHI 2HU, England.

